

Theoretical and Quantitative Structural Relationship Study of the Electrochemical Properties of $[M-2@C-x]@[SWCNT(5,5)\text{-Armchair-C}_n\text{H}_{20}]$ ($M = \text{Er and Sc}$, $x=82$ and 84 , and $n=20\text{-}300$) Complexes

TAHERPOUR, Avat Arman

Univ Queensland, Sch Mol & Microbial Sci, Unusual Mol & React Intermediates Grp, Chem Bldg, Brisbane, Qld 4072 Australia
&
Islam Azad Univ, Fac Sci, Dept Chem, Arak, Iran
avatarman.taherpour@gmail.com

Abstract. One of the nanoscale structures of carbon is the carbon nanotubes. These structures of carbon display an attractive variation of structural characteristics, and many useful forms have been synthesized and identified. Carbon nanotubes are either single-wall (SWCNT) or multiwall (MWCNT); the former attract attention due to their unique electronic, optical, and spectroscopic properties. One of the main recognized structures of single-walled nanotubes is the (5,5) single-walled tube. Endohedral metallofullerenes ($M-n@C-x$) were introduced as a new class of spherical fullerenes group with unique properties. Formation of endohedral metallofullerenes is thought to involve the transfer of electrons from the encapsulated metal atom(s) to the surrounding fullerene cage. Two of these molecules are $\text{Er-}2@C\text{-}82$ (1) and $\text{Sc-}2@C\text{-}84$ (2). A topological index is a mathematical invariant of a chemical graph, which shows a significant correlation with some chemical or physical properties. Topological indices have been successfully used to construct effective and useful mathematical methods for finding good relationships between structural data and the properties of these materials. To establish a good structural relationship between the structure of molecules $\text{Sc-}2@C\text{-}84$, $\text{Er-}2@C\text{-}82$, and $[SWCNT(5,5)\text{-Armchair-C}_n\text{H}_{20}]$ ($n = 20\text{-}190$) 320, the molecular degree of unsaturation (D_u) was used as one of the useful numerical and structural properties of unsaturated compounds. In this study, the relationship between this index and the free energy of electron transfer ($\Delta G(\text{et})$) as assessed using the Rehm-Weller equation on the basis of the first oxidation potential ($E\text{-ox}(1)$) of $\text{Sc-}2@C\text{-}84$ and $\text{Er-}2@C\text{-}82$ for the predicted supramolecular complexes between 3-20 and the endohedral metallofullerenes $\text{Sc-}2@C\text{-}84$ and $\text{Er-}2@C\text{-}82$ as $[M-2@C-x]@[SWCNT(5,5)\text{-Armchair-C}_n\text{H}_{20}]$ ($M = \text{Er and Sc}$, $x = 82$ and 84 , and $n = 20\text{-}190$) 21-38 and 39-56 are presented. The results were extended for $[M-2@C-x]@[SWCNT(5,5)\text{-Armchair-C}_n\text{H}_{20}]$ ($M = \text{Er and Sc}$, $x = 82$ and 84 , and $n = 200\text{-}300$) 68-78 and 79-89.

Keywords. Wall carbon nanotubes; Density-functional thermochemistry; Endohedral metallofullerene; Electronic-structure; Topological indexes; Exact exchange; Fullerenes; C-60; Molecules; Isomers

References

1. ***, PHYS WORLD 13; 2000
2. ANDERSON MR. Making connections between metallofullerenes and fullerenes: electrochemical investigations. CARBON 38:1663; 2000
3. ARELLANO JS. Interaction of molecular and atomic hydrogen with (5,5) and (6,6) single-wall carbon nanotubes. JOURNAL OF CHEMICAL PHYSICS 117:2281; 2002
4. AVOURIS P. Molecular electronics with carbon nanotubes. ACCOUNTS OF CHEMICAL RESEARCH 35:1026; 2002
5. BANDOW S. Smallest limit of tube diameters for encasing of particular fullerenes determined by radial breathing mode Raman scattering. CHEMICAL PHYSICS LETTERS 347:23; 2001
6. BARNETT R. Superconducting and charge-density wave instabilities in ultrasmall-radius carbon nanotubes. SOLID STATE COMMUNICATIONS 135:335; 2005
7. BECKE AD. DENSITY-FUNCTIONAL THERMOCHEMISTRY. 3. THE ROLE OF EXACT EXCHANGE. JOURNAL OF CHEMICAL PHYSICS 98:5648; 1993
8. BOLBOACA SD. How good can the characteristic polynomial be for correlations? INTERNATIONAL JOURNAL OF MOLECULAR SCIENCES 8:335; 2007
9. BRUS LE. A SIMPLE-MODEL FOR THE IONIZATION-POTENTIAL, ELECTRON-AFFINITY, AND AQUEOUS REDOX POTENTIALS OF SMALL SEMICONDUCTOR CRYSTALLITES. JOURNAL OF CHEMICAL PHYSICS 79:5566; 1983
10. COLLINS PG. SCI AM 283:38; 2000
11. DENNIS TJS. Isolation and characterisation of the two major isomers of [84]fullerene (C-84). CHEMICAL COMMUNICATIONS :619; 1998
12. DRESSSELHAUS MS. CARBON-FIBERS BASED ON C-60 AND THEIR SYMMETRY. PHYSICAL REVIEW B 45:6234; 1992
13. DRESSSELHAUS MS. SCI FULLERENES CARBO. 1996
14. DRESSSELHAUS SM. CARBON NANOTUBES SYN. 2001
15. DU YP. J CHEM INF COMP SCI 42:1128; 2002
16. FOWLER PW. ATLAS FULLERENES 30; 1995
17. FUCHS D. Extraction and chromatographic elution behavior of endohedral metallofullerenes: Inferences regarding effective dipole moments. JOURNAL OF PHYSICAL CHEMISTRY 100:725; 1996
18. HANG LY. J PHYS CHEM-US 99:16479; 1995
19. HANSEN PJ. CHEMICAL APPLICATIONS OF GRAPH-THEORY. 1. FUNDAMENTALS AND TOPOLOGICAL INDEXES. JOURNAL OF CHEMICAL EDUCATION 65:574; 1988
20. HAUFLER RE. EFFICIENT PRODUCTION OF C60 (BUCKMINSTERFULLERENE), C60H36, AND THE SOLVATED BUCKIDE ION. JOURNAL OF PHYSICAL CHEMISTRY 94:8634; 1990
21. HOFFMAN KR. Spectroscopic studies of fullerenes doped with rare earth and transition metal ions. JOURNAL OF LUMINESCENCE 66:244; 1995
22. HOSOYA H. TOPOLOGICAL INDEX - NEWLY PROPOSED QUANTITY CHARACTERIZING TOPOLOGICAL NATURE OF STRUCTURAL ISOMERS OF SATURATED HYDROCARBONS. BULLETIN OF THE CHEMICAL SOCIETY OF JAPAN 44:2332; 1971
23. IIDUKA Y. CHEM COMMUN 2057:2059; 2006
24. ITO Y. Enhanced 1520 nm photoluminescence from Er³⁺ ions in di-erbium-carbide metallofullerenes (Er₂C₂)@C-82 (isomers I, II, and III). ACS NANO 1:456; 2007
25. JANDA P. Nanostructuring of highly ordered C-60 films by charge transfer. ADVANCED MATERIALS 10:1434; 1998

26. JEHOULET C. ELECTROCHEMISTRY AND LANGMUIR TROUGH STUDIES OF C-60 AND C-70 FILMS. JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 114:4237; 1992
27. KAVAN L. Electrochemical tuning of electronic structure of carbon nanotubes and fullerene peapods. CARBON 42:1011; 2004
28. KIKUCHI K. CHARACTERIZATION OF THE ISOLATED Y-AT-C-82. JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 116:9367; 1994
29. KIMURA K. Evidence for substantial interaction between Gd ion and SWNT in (Gd@C-82)(n)@SWNT peapods revealed by STM studies. CHEMICAL PHYSICS LETTERS 379:340; 2003
30. KUDIN KN; Hybrid density-functional theory and the insulating gap of UO₂. PHYSICAL REVIEW LETTERS 89:ARTN266402; 2002
31. LOURIE O. Buckling and collapse of embedded carbon nanotubes. PHYSICAL REVIEW LETTERS 81:1638; 1998
32. MAKOV G. ON THE IONIZATION-POTENTIAL OF SMALL METAL AND DIELECTRIC PARTICLES. JOURNAL OF CHEMICAL PHYSICS 88:5076; 1988
33. MINKIN VI. Glossary of terms used in theoretical organic chemistry (IUPAC recommendations 1999). PURE AND APPLIED CHEMISTRY 71:1919; 1999
34. MINTMIRE JW. ARE FULLERENE TUBULES METALLIC. PHYSICAL REVIEW LETTERS 68:631; 1992
35. MIYAKE T. Electronic structureOf C-60-encapsulating semiconducting carbon nanotube. SOLID STATE COMMUNICATIONS 125:201; 2003
36. MUSCAT J. On the prediction of band gaps from hybrid functional theory. CHEMICAL PHYSICS LETTERS 342:397 2001
37. NAGASE S. THEORETICAL-STUDY OF THE DIMETALLOFULLERENE SC-2-AT-C-84. CHEMICAL PHYSICS LETTERS 231:319; 1994
38. OKIMOTO H. Element-specific magnetic properties of di-erbium Er-2@C-82 and Er₂C₂@C-82 metallofullerenes: A synchrotron soft X-ray magnetic circular dichroism study. JOURNAL OF PHYSICAL CHEMISTRY C 112:6103; 2008
39. PERRY JK. Ab initio evidence for the formation of impurity d(3z²-2)(r) holes in doped La_{2-x}Sr_xCuO₄. PHYSICAL REVIEW B 65:ARTN144501; 2002
40. PICHLER T. Proof for trivalent Sc ions in Sc-2 @ C-84 from high-energy spectroscopy. PHYSICAL REVIEW B 62:13196; 2000
41. RAGHAVACHARI K. Perspective on "Density functional thermochemistry. III. The role of exact exchange" - Becke AD (1993) J Chem Phys 98:5648-52. THEORETICAL CHEMISTRY ACCOUNTS 103:361; 2000
42. RANDIC M. On characterization of molecular attributes. ACTA CHIMICA SLOVENICA 45:239; 1998
43. RANDIC M. CHARACTERIZATION OF MOLECULAR BRANCHING. JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 97:6609; 1975
44. REHM D. KINETICS OF FLUORESCENCE QUENCHING BY ELECTRON AND H-ATOM TRANSFER. ISRAEL JOURNAL OF CHEMISTRY 8:259; 1970
45. RUCKER G. On topological indices, boiling points, and cycloalkanes. JOURNAL OF CHEMICAL INFORMATION AND COMPUTER SCIENCES 39:788; 1999
46. RUOFF RS. THE RELATIONSHIP BETWEEN THE ELECTRON-AFFINITIES AND HALF-WAVE REDUCTION POTENTIALS OF FULLERENES, AROMATIC-HYDROCARBONS, AND METAL-COMPLEXES. JOURNAL OF PHYSICAL CHEMISTRY 99:8843; 1995
47. SHEN HJ. The compressive mechanical properties of C-n (n=20, 60, 80, 180) and endohedral M@C-60 (M = Na, Al, Fe) fullerene molecules. MOLECULAR PHYSICS 105:2405; 2007

48. SHERIGARA BS. Electrocatalytic properties and sensor applications of fullerenes and carbon nanotubes. *ELECTROANALYSIS* 15:753; 2003
49. SLANINA Z. Ca@C-82 isomers: Computed temperature dependency of relative concentrations. *JOURNAL OF CHEMICAL PHYSICS* 120:3397; 2004
50. SLANINA Z. *MATCH COMMUN MATH CO* 44:335; 2001
51. SMALLEY RE. DOPING THE FULLERENES. *ACS SYMPOSIUM SERIES* 481:141; 1992
52. SMITH BW. *NATURE* 396:3239; 1998
53. SRIVASTAVA D. Computational nanotechnology with carbon nanotubes and fullerenes. *COMPUTING IN SCIENCE & ENGINEERING* 3:42; 2001
54. SRIVASTAVA D. *P IEEE SUP*:97; 1997
55. SRIVASTAVA D. Nanoplasticity of single-wall carbon nanotubes under uniaxial compression. *PHYSICAL REVIEW LETTERS* 83:2973; 1999
56. STEVENSON S. AUTOMATED HPLC SEPARATION OF ENDOHEDRAL METALLOFULLERENE SC-AT-C-2N AND Y-AT-C-2N FRACTIONS. *ANALYTICAL CHEMISTRY* 66:2675; 1994
57. SUENAGA K. Direct imaging of Sc-2@C-84 molecules encapsulated inside single-wall carbon nanotubes by high resolution electron microscopy with atomic sensitivity. *PHYSICAL REVIEW LETTERS* 90:ARTN055506; 2003
58. SUZUKI T. ELECTROCHEMICAL PROPERTIES OF LA-AT-C-82. *JOURNAL OF THE AMERICAN CHEMICAL SOCIETY* 115:11006; 1993
59. TAHERPOUR A. Quantitative relationship study of mechanical structure properties of empty fullerenes. *FULLERENES NANOTUBES AND CARBON NANOSTRUCTURES* 16:196; 2008
60. TAHERPOUR A. The structural relationship between Randic indices, adjacency matrixes, distance matrixes and maximum wave length of linear simple conjugated polyene compounds. *JOURNAL OF MOLECULAR STRUCTURE-THEOCHEM* 726:183; 2005
61. TAHERPOUR AA. *2 INT C CHEM ED SUST*; 2004
62. TAHERPOUR AA. *AUST J ED CHEM* 65:37; 2005
63. TAHERPOUR AA. Structural relationship between degree of unsaturation with polarizability of (5,5) armchair single-wall carbon nanotubes. *FULLERENES NANOTUBES AND CARBON NANOSTRUCTURES* 15:279; 2007
64. TAHERPOUR AA. *NANOTUBES CARBON NAN* 17:26; 2009
65. TOPOL IA. Experimental determination and calculations of redox potential descriptors of compounds directed against retroviral zinc fingers: Implications for rational drug design. *PROTEIN SCIENCE* 10:1434; 2001
66. TOUZIK A. Nanostructuring of potassium fulleride layers. *EUROPHYSICS LETTERS* 60:411; 2002
67. WANG CR. A scandium carbide endohedral metallofullerene: (Sc₂C₂)@C-84. *ANGEWANDTE CHEMIE-INTERNATIONAL EDITION* 40:397; 2001
68. WEAVER JH. XPS PROBES OF CARBON-CAGED METALS. *CHEMICAL PHYSICS LETTERS* 190:460; 1992
69. WIENER H. *J AM CHEM SOC* 17:20; 1947
70. XIE QS. ELECTROCHEMICAL DETECTION OF C-60(6-) AND C-70(6-) - ENHANCED STABILITY OF FULLERIDES IN SOLUTION. *JOURNAL OF THE AMERICAN CHEMICAL SOCIETY* 114:3978; 1992
71. YAKOBSON BI. *MECH PROPERTIES CARB* :293; 2001
72. YAKOBSON BI. Nanomechanics of carbon tubes: Instabilities beyond linear response. *PHYSICAL REVIEW LETTERS* 76:2511; 1996
73. YANNONI CS. SCANDIUM CLUSTERS IN FULLERENE CAGES. *SCIENCE* 256:1191; 1992

74. ZHANG M. Thermogravimetric analysis for the array of C-60 molecules formed in single-wall carbon nanotube. CHEMICAL PHYSICS LETTERS 369:680; 2003
75. ZHOU ZY. Electronic structure of tubular aromatic molecules derived from the metallic (5,5) armchair single wall carbon nanotube. JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 126:3597; 2004