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Modeling the Octanol-Water Partition Coefficient of Substituted Phenols by the Use of Structure Information

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Aim: The paper presents the abilities in estimation and prediction of octanol-water partition coefficient of some para-substituted phenols through integration of complex structure information by the use of an original molecular descriptors family on structureproperty relationships.

Material and Method: The measured octanol-water partition coefficient of thirty parasubstituted phenols, express in logarithmical scale was taken from a previous study [¹]. All para-substituted phenols were used in order to generate and calculate the molecular descriptors family [²]. Based on the generated descriptors, the structure-property relationships models were build. The obtained models (bi-, and four-varied) were validated through the assessment of the cross-validation leave-one-out score, and analyzed through the squared correlation coefficient and the model stability. The comparison between the bi and four-varied models was performed by applying the Steiger's Z test.

Results: The characteristics of the bi-varied model are: the squared correlation coefficient (r^2) equal with 0.8943, the leave-one-out scores (r^2_{cv-loo}) equal with 0.8659, and the model stability (r^2 - r^2_{cv-loo}) equal with 0.02838. The multiple linear regression analysis shows that best results are obtained in four-varied model (\hat{Y} = 8.69·10⁻²+5.56·10⁻³·*isDDkGg*-0.42·*IMmrKQg*+9.41·10⁻³·*IPMDKQg*-7.8·10⁻²·*IFMMKQg*): r^2 = 0.9781, proving to be a stable model (r^2_{cv-loo} = 0.01) and having best prediction ability (r^2_{cv-loo} = 0.9680). The correlation coefficient obtained by the four-varied model is statistically significant higher comparing with the correlation coefficient obtained by the bi-varied model (Steiger's Z parameter = 4.3501, p-value = 6.81·10⁻⁶).

Conclusion: The four-varied model indicates that the octanol-water partition coefficient of studied para-substituted phenols is like to be of geometry nature, is strongly dependent on the partial change of compounds and group electronegativity and it is in relation with elastic force.

 $^[^{1}]$ T.W. Schultz, Relative toxicity of para-substituted phenols: Log K_{OW} and pKa-dependent structure-activity relationships. Bull. Environ. Contam. Toxicol., 1987, 38, 994-9.

^{[&}lt;sup>2</sup>] L. Jäntschi, Molecular Descriptors Family on Structure Activity Relationships 1. The review of Methodology, Leonardo Electronic Journal of Practices and Technologies, 2005, 6, 76-98.

ODELING THE OCTANOL-WATER PARTITION COEFFICIENT OF SUBSTITUTED PHENOLS BY THE USE OF STRUCTURE INFORMATION

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The poster presents the abilities in estimation and prediction of octanol-water partition coefficient of some para-substituted phenols through integration of complex structure information by the use of an original molecular descriptors family on structure-property relationships.

MATERIALSAND METHODS

The measured octanol-water partition coefficient of thirty para-substituted phenols, express in logarithmical scale was taken from a previous study [1]. All para-substituted phenols were used in order to generate and calculate the molecular descriptors family [2]. Based on the generated descriptors, the structure-property relationships models were build. The obtained models (bi-, and four-varied) were validated through the assessment of the crossvalidation leave-one-out score, and analyzed through the squared correlation coefficient and the model stability. The comparison between the bi and four-varied models was performed by applying the Steiger's Z test.

RESULTS

Two MDF-SAR models, one bi- and one four-varied, proved to have abilities in estimation and prediction of octanol-water partition coefficient for parasubstituted phenols. The MDF-SAR models are: (1)

• Bi-varied: $Y_{2y} = 1.07 + 3.38 \cdot 10^{-3}$ isDDkGg - 0.40 IMmrKQg

(2)Four-varied: $Y_{4v} = 8.69 \cdot 10^{-2} + 5.56 \cdot 10^{-3}$ is DbkGg - 4.16 $\cdot 10^{-1}$ /MmrKQg + 9.41 $\cdot 10^{-3}$ /PMDKQg - 7.80 $\cdot 10^{-2}$ /FMMKQg Statistical characteristics of bi- and four-varied MDF-SAR models are in Table 1 The plot of the estimated log Kow by four-varied MDF-SAR model versus

measured log K_{ow} is in Figure 1.

Table 1. Statistical characteristics of the MDF-SAR models

	Value			
Parameter	Bi -varied	Four -varied		
	(n = 30, v = 2)	(n = 30, v = 4)		
r (correlation coefficient)	0.9457	0.9890		
Cl[lower, upper] (95% confidence intervals)	[0.8897, 0.9740]	[0.9767, 0.9948]		
r ² (squared correlation coefficient)	0.8943	0.9781		
r ² adjusted correlation coefficient)	0.8865	0.9745		
s at (standard error)	0.3671	0.1739		
F == (Fisher parameter of regression model)	114	279		
peet (F significance)	< 0.0001	< 0.000		
r ² _{cv400} (cross -validation loo squared correlation coefficient)	0.8660	0.9680		
s to (standard error of cross -validation loo analysis)	0.4139	0.2101		
F _{pred} (Fisher p arameter in loo analysis)	87	189		
pred (%) (F significance in loo analysis)	< 0.0001	< 0.000		
r ² - r ² _{cvko} (measure of model stability)	0.028 4	0.0100		
r²(isDDkGg , IMmrKQg) [.]	0.0760	0.0760		
r²(isDDkGg , IPMDKQg) [.]	n.a.	0.3346		
r²(isDDkGg , IFMMKQg)*	n.a.	0.0083		
r²(IMmrKQg , IPMDKQg) [,]	n.a.	0.0232		
r²(IMmrKQg , IFMMKQg) [,]	n.a.	0.1624		
r²(IPMDKQg , IFMMKQg)"	n.a.	0.6214		
n = number of compounds, v = number of descriptor, * -sauared correlation	n coefficient between descripto	rs , n.a. = not applicabl		

Figure 1. Estimated versus measured log Kow



The correlation coefficient obtained by the four-varied model is statistically significant higher comparing with the correlation coefficient obtained by the bi-varied model (Steiger's Z parameter = 4.3501, p-value = $6.81 \cdot 10^{-6}$).

Validation of the four-varied MDF-SAR model was performed by spilling the set of para-substituted phenols in training and test sets. The characteristics of the regression models and theirs performances in training and test sets are in Table 6. There were included into table 2.

The abilities of four-varied MDF-SAR model in estimation and prediction obtained in training versus test analysis, when the number of compounds in training set was equal with 2/3 from the total number of compounds is in Figure 2. Figure 2. Prediction and estimation abilities of four-varied

Training sets			Test se ts				7	
No	r #	95%Cl r #	F∗	Nots	۲ъ	95%Cl r ₅	F ts	∠ rtr⊸rts
15	0.9958	[0.9870, 0.9986]	299 *	15	0.9808	[0.9416, 0.9937]	52.	1.871 .
16	0.9891	[0.9679, 0.9963]	124 ·	14	0.9855	[0.9534, 0.9955]	35.	1.197
17	0.9858	[0.9600, 0.9949]	104 ·	13	0.9880	[0.9591, 0.9965]	80.	0.630
18	0.9877	[0.9665, 0.9955]	130.	12	0.9875	[0.9545, 0.9965]	43.	0.019
19	0.9922	[0.9793, 0.9970]	221 .	11	0.9807	[0.9249, 0.9951]	29.	1.053
20	0.9901	[0.9745, 0.9961]	186 .	10	0.9851	[0.9360, 0.9965]	39.	0.458
21	0.9918	[0.9794, 0.9967]	241 .	9	0.9834	[0.9203, 0.9966]	20	0.753
22	0.9856	[0.9649, 0.9941]	145 .	8	0.9921	[0.9552, 0.9986]	34	0.600
23	0.9881	[0.9716, 0.9950]	186 .	7	0.9882	[0.9191, 0.9983]	20	0.007
a = 0.0001 $a = 0.0001$ $a = 0.0001$ $a = 0.005$ $a = 0.005$								

Table 2. Statistical characteristics of models in training versus test analysis



MDF-SAR model in training versus test analysis

CONCLUSIONS

Octanol-water partition coefficient of para-substituted phenols, being a property of chemicals, is proved to be related with complex information obtained from the compounds structure, being strongly related with compounds geometry, theirs partial change and elastic force, in relationship with group electronegativity and inverse related with the property potential.

The goodnes-of-fit of the four-varied MDF-SAR model and internal validation results sustain that the model is stable and valid. Future studies on new external para-substituted phenols are necessary in order to assess the four-varied model robustness and predictivity.

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[1] T.W. SCHULTZ, Relative toxicity of para-substituted phenols: Log K_{aw} and pKa-dependent structure-activity relationships. Bull. Environ. Contam. Toxicol., 1987, 38, 994-9. [2] L. JÄNTSCHI, Molecular Descriptors Family on Structure Activity Relationships 1. The review of Methodology, Leonardo Electronic Journal of Practices and Technologies, 2005, 6, 76-98